Enantiomerically Pure α,β -Unsaturated Five-Membered-Ring Aldehydes by Ring Contraction of Epoxyhexopyranosides

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A series of epoxyhexopyranosides, variously substituted in the 6-position, were each transformed by ring contraction into a single, enantiomerically pure, α,β -unsaturated furanosidic aldehyde. Similar ring contraction of a C-propylglycosidic analog of one of the epoxyhexopyranosides gave a mixture of two diastereomeric aldehydes. This finding supports the previously suggested mechanism of the reaction and indicates that O-glycosidic epoxypyranosides also rearrange into two aldehydes, one of which is unstable.

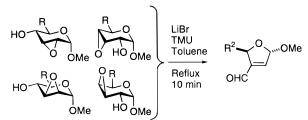
Introduction

Ring contraction of epoxycyclohexanes and epoxycyclohexanols provides cyclopentane- and cyclopentenecarboxaldehydes, respectively.¹⁻⁴ The latter have been used as starting materials for the syntheses of a number of terpenes. Similarly, epoxypyranosides give α,β -unsaturated furanosidic aldehydes, 6,7 which were used for the synthesis of enantiomerically pure tetrahydrofuran-based natural products, including several lignans.8 The mechanism of the ring contraction has been investigated using deuterium-labeled epoxy alcohols.^{4,6,7}

Epoxycyclohexanols^{3,4} normally give high yields (>80%) of the ring-contracted aldehydes as isomeric mixtures that are difficult to separate into the individual aldehydes. In contrast, epoxypyranosides give only a single aldehyde that is easily purified, but the yields are modest (~25-60%).^{6,7} However, different stereo- and regioisomers of the epoxypyranosides provide the same aldehyde (cf. Scheme 1), which makes it possible to use readily available mixtures of sugar epoxides for the ring contraction. It should also be noted that the methyl *O*-glycosidic aldehydes can undergo a 1,4-elimination of methanol, thus providing a route to substituted furan-3-aldehydes.^{6,7} Ring contractions have also been performed with pyranosidic sulfonyl esters, which gave some furanosidic saturated and α,β -unsaturated aldehydes. ^{9,10}

In order to investigate the generality of the formation of a single stable aldehyde from epoxypyranosides, we submitted a number of these compounds to ring contrac-

Scheme 1



tion (Scheme 2). A C-glycoside analog of one of the O-glycosidic epoxypyranosides was also synthesized and ring-contracted, which gave a mixture of two isomeric aldehydes (Scheme 4). This experiment provided an explanation for the fortuitous formation of a single aldehyde in most ring contractions of O-glycosidic epoxypyranosides.

Results and Discussion

I. Ring Contraction of Methyl Epoxyhexopyra**nosides.** Treatment of the epoxy alcohols 11,12 **1–6** with LiBr and tetramethylurea (TMU) in refluxing toluene for 10 min resulted in ring contraction and dehydration, which produced the α,β -unsaturated aldehydes 10-15 in 57–38% yield (Scheme 2). Only one aldehyde could be isolated from each reaction mixture, except for 15, which contained approximately 3% of an isomeric aldehyde (**15a**). The latter was not obtained in pure form, but its ¹H NMR spectrum (mixture of **15** and **15a**) was consistent with a structure corresponding to the unstable aldehyde depicted in Scheme 3. The chromatographic isolation of **10−15** on silica gel was simple since they all had a higher R_f value than the byproducts.

Sulfone epoxide 7 gave the aldehyde 16 in only 9% yield, and the sulfone epoxides 8 and 9 gave no aldehyde product at all. Compound 8 was recovered (75%) after standard treatment with LiBr/TMU, and compound 9 was only consumed after prolonged reaction in the presence of additional LiBr. It seems as if the sulfone group competes with the epoxide oxygen for lithium ion, thus interfering with bromide ion attack on the epoxide carbon and consequently reducing the amount of the bromohydrin intermediate available for ring contraction. A simplified reaction mechanism is depicted in Scheme 3; for a full discussion, see refs 4, 6, and 7.

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⁽¹⁾ Rickborn, B.; Gerkin, R. M. J. Am. Chem. Soc. 1971, 93, 1693.

⁽²⁾ Magnusson, G. Org. Prep. Proc. Int. 1990, 22, 547.

⁽²⁾ Magnusson, G.; Thorén, S. *J. Org. Chem.* **1973**, *38*, 1380. (4) Bergman, R.; Magnusson, G. *J. Org. Chem.* **1986**, *51*, 212. (5) Lactaral: Froborg, J.; Magnusson, G.; Thorén, S. *Acta Chem. Scand. Ser. B* **1974**, *28*, 265. Tanis, S. P.; Head, D. B. *Tetrahedron* Lett. 1982, 23, 5509. Marasmic acid: Greenlee, W. J.; Woodward, R. B. Tetrahedron 1980, 36, 3367, 3361. Hirsutene: Hudlicky, T.; Kutchan, T. M.; Wilson, S. R.; Mao, D. T. J. Am. Chem. Soc. 1980, 102, 6351. Capnellene: Stevens, K. E.; Paquette, L. A. Tetrahedron Lett. 1981, 22, 4393. Silphinene: Paquette, L. A.; Leone-Bay, A. J. Am. Chem. Soc. 1983, 105, 7352. (+)-Isovelleral: Bergman, R.; Hansson, T.; Sterner, O.; Wickberg, B. J. Chem. Soc., Chem. Commun. 1990, 865.

⁽⁶⁾ Sundin, A.; Frejd, T.; Magnusson, G. J. Org. Chem. 1986, 51,

⁽⁷⁾ Rehnberg, N.; Magnusson, G. *J. Org. Chem.* **1990**, *55*, 5467. (8) Botryodiplodin: Rehnberg, N.; Magnusson, G. *Acta Chem. Scand.* **1990**, *44*, 377. Lignans: Rehnberg, N.; Magnusson, G. *J. Org. Chem.* **1990**. 55, 4340.

⁽⁹⁾ Binkley, R. W. J. Org. Chem. 1992, 57, 2353.

⁽¹⁰⁾ Kassou, M.; Castillón, S. J. Org. Chem. 1995, 60, 4353.

⁽¹¹⁾ Pontén, F.; Magnusson, G. Acta Chem. Scand. 1994, 48, 566.(12) Pontén, F.; Magnusson, G. J. Org. Chem. 1996, 61, 7463.

^a (a) LiBr, TMU, toluene, reflux, 10 min; (b) 3% of the isomer 15a (see Experimental Section) was also formed; (c) no aldehyde was obtained.

The ease of preparation and the purity of the aldehydes shown in Scheme 2 makes the ring contraction a useful preparative procedure. However, it is disturbing not knowing the reasons for the modest yields obtained. On the basis of mechanistic investigations with deuterated epoxycyclohexanols⁴ and epoxypyranosides, ^{6,7} we expect that not only the isolated aldehyde is formed in the ring contraction but also its isomer (cf. the unstable aldehyde of Scheme 3). A very limited number of α,β -unsaturated aldehydes carrying an (RO)₂CH group in the α-position is known. They are generally difficult to prepare, and more importantly, they are prone to undergo elimination of alcohol from the acetal moiety, as well as polymerization.¹³ Such side reactions would explain the limited yield of isolated aldehydes (Scheme 2), as well as the absence of isomers (except for 15a).

II. Synthesis and Ring Contraction of C-Glycosidic Epoxy Alcohols. In order to investigate the importance of the methoxy group for the formation of

Scheme 3a

^a Simplified mechanistic scheme; for a full account, see refs 4, 6, and 7.

byproducts, the *C*-epoxyglycoside analogs **24** and **25** were prepared and submitted to ring contraction (Scheme 4).

Allyl C-glycoside 17^{14} was hydrogenated to give the C-propyl analog 18 (95%). Deacetylation of 18 with methanolic MeONa, followed by acetalization with α,αdimethoxytoluene gave the 4,6-O-benzylidene acetal 19 (88%). Treatment of **19** with *p*-TsCl-K₂CO₃-NaOH-DMSO in toluene¹⁵ gave the corresponding ditosylate, which was treated, without purification, with the twophase system H₂O-toluene-NaOH-Bu₄NHSO₄-MeO-CH₂CH₂OH-DMSO to give the epoxide **20** (18%). In an attempt to raise the yield, 19 was tosylated in pyridine, which gave a mixture of the two corresponding monotosylates. Treatment of the mixture with methanolic MeONa gave the epoxides 21 (65%) and 20 (25%). The benzylidene group of 20 and 21 was removed by hydrogenolysis over several days, which gave 22 (74%) and 23 (75%), respectively. Silylation of 22 with dimethylhexylchlorosilane gave the epoxy alcohol 24 (16%) after 3 days. The low yield was probably due to migration of the silyl group from O-6 to O-4, followed by silvlation of O-6, since a significant amount of disilylated material was also obtained. Silylation of 23 under the same conditions gave, after 18 h, epoxy alcohol 25 (61%), as well as some disilylated material. The sluggishness of the epoxidation, hydrogenolysis, and silvlation reactions was unexpected, since the O-glycoside analogs reacted readily under the same conditions.^{7,15}

Ring contraction (LiBr, TMU) of the epoxy alcohols 24 and 25 gave mixtures of the two aldehydes 26 and 27, together with the ketone 28. No other products could be identified. The ketone was probably formed via a 1,2hydride shift followed by loss of water. Such hydride shifts were observed in the rearrangement of epoxycyclohexanols described previously.⁴ The total product yield in the rearrangement of 24 and 25 was lower than expected from the TLC analysis. It might in part be due

^{(13) (}a) Nishino, T.; Miichi, Y.; Tokyuama, K. Bull. Chem. Soc. Jpn. **1973**, *46*, 580. (b) Tanaka, M.; Abe, Y.; Tokuama, K. *Chem. Pharm. Bull.* **1978**, *26*, 1558. (c) Depezay, J. C.; Merrer, Y. L.; Saniere, M. Synthesis 1985, 8, 766. (d) Boussoufi, A.; Parrain, J.-L.; Hudhomme, P.; Daguay, G. Tetrahedron 1994, 50, 12609.

^{(14) (}a) Bennek, J.; Gray, G. J. Org. Chem. 1987, 52, 892. (b) Horton, D.; Miyake, T. Carbohydr. Res. 1988, 184, 221.
(15) Szeja, W. Carbohydr. Res. 1988, 183, 135.

 a (a) $H_2,$ Pd/C, EtOAc/toluene 1:1, 5 h, 22 °C; (b) MeONa/MeOH, 3 h, 22 °C, then (MeO) $_2 CHC_6 H_5,$ p-TsOH, MeCN, 24 h, 22 °C; (c) p-TsCl, $K_2 CO_3,$ NaOH, toluene/DMSO, 22:1, 22 h, 22 → 55 °C, then MeOCH $_2 CH_2 OH,$ Bu $_4 NHSO_4,$ 28 h, 22 °C (→ ditosylate), then 50% aqueous NaOH, Bu $_4 NHSO_4,$ DMSO, MeOCH $_2 CH_2 OH,$ toluene, 8.7 d, 22 °C → reflux; (d) p-TsCl, pyridine, 24 h, 60 °C, then MeONa/MeOH, 96 h, 22 °C; (e) $H_2,$ Pd/C, EtOAc/THF 1:3, 6 d (20) and 10 d (23), 22 °C; (f) Me $_2 CHMe_2 CMe_2 SiCl,$ pyridine, 72 h (24) and 18 h (25), 22 °C; (g) LiBr, TMU, toluene, reflux, 10 min.

to losses in the workup, since the products are quite volatile.

The aldehydes **26** and **27** could not be separated by preparative column chromatography. Attempted separation with HPLC, both on SiO_2 and C_8 columns, was equally unsuccessful. Similar difficulties would probably have arisen with the O-glycosidic aldehydes of Scheme 2, if the unstable isomeric aldehyde (cf. Scheme 3) had not been destroyed in the rearrangement reaction. This was corroborated by unsuccessful attempts to separate **15** from its isomer **15a**.

Aldehyde **26** was obtained, by preparative GC, in acceptable purity (90–95%) for structure determination by NMR (¹H–¹H NOESY and ¹H–¹³C long-range HET-COR). Thus, H-4 showed a strong NOE with CHO, H-5,

H-1", Me₂Si, and Me₂CH, but not with H-2 and H-1' (for numbering, see Scheme 4). In addition, the long-range HETCOR experiment showed couplings between C-2 and CHO, and between C-4 and H-1". The NMR spectra of **27** (in a mixture with **26**) were fully consistent with the given structure. The observation that aldehyde **26** was formed from the C-glycosidic epoxy alcohol **24** indicates that an isomer of **10** was probably also formed in the reaction of the analogous O-glycosidic epoxy alcohol **1**, but it was destroyed during the reaction, thus greatly simplifying the isolation of **10**.

Experimental Section

¹H NMR spectra were recorded (23 °C) at 400 or 300 MHz proton frequency, using CDCl₃ or C₆D₆ as solvent and CHCl₃ (δ 7.26 ppm) or C₆D₅H (δ 7.30 ppm) as internal standards. ¹³C NMR spectra were recorded at 100 or 75 MHz carbon frequency, using the same solvents and internal standards (δ 77.0 and 128.0 ppm, respectively) as above. The compounds described below with signal assignments were investigated by 2D NMR experiments. LiBr was dried under reduced pressure (0.1 mmHg) at 120 °C for 2-12 h. Tetramethylurea (TMU) was distilled and kept over 4 Å molecular sieves. TLC analyses were performed with Merck SiO_2 60 F_{254} precoated aluminum sheets with visualization by UV light, by I2, or by charring with anisaldehyde in ethanolic sulfuric acid. 16 Preparative column chromatography was performed with Matrex SiO₂ 60 (35–70 mm) unless otherwise stated. Compounds 1,⁷ $3.^{12}$ **4**, 11 **5**, 12 **6**, 12 **7**, 11 **8**, 11 **9**, 11 **10**, 7 and 17 have been described.

Methyl 2,3-Anhydro-6-deoxy-6-propyl-α-D-allopyranoside (2). Methyl 2,3-anhydro-4-benzoyl-6-deoxy-6-(prop-2-en-1-yl)- α -D-allopyranoside 17 (378 mg, 1.24 mmol) was dissolved in EtOAc (30 mL), and the mixture was hydrogenated (H₂, 1 atm, Pd/C, 74 mg) for 15 h. The catalyst was filtered off (Celite), and the solvent was removed. The crude product was dissolved in MeOH (10 mL) and debenzoylated by treatment with methanolic NaOMe (0.25 mL, 0.5 M) for 20 h. The mixture was neutralized with SiO2 and concentrated. The residue was chromatographed (heptane/EtOAc, 1:1) to give 2 (198 mg, 79%) as an oil: $[\alpha]^{23}$ _D +167 (c 0.9, CHCl₃); ¹H NMR data ($\overline{C_6D_6}$) δ 4.57 (d, 1 H, J = 2.7 Hz), 3.79 (dt, 1 H, J = 2.3, 9.1 Hz), 3.49 (br dd, 1 H, J = 8.6, 9.1 Hz), 3.35 (s, 3 H), 3.14 (dd, 1 H, J = 2.8, 4.1 Hz), 3.12 (dd, 1 H, J = 1.5, 4.1 Hz), 2.01(m, 1 H), 1.94 (br d, 1 H, J = 8.9 Hz), 1.70 (m, 1 H), 1.34–1.60 (m, 4 H), 1.04 (t, 3 H, J = 7.1 Hz); ¹³C NMR data (CDCl₃) δ 94.5, 69.9, 68.7, 56.0, 55.7, 54.1, 31.1, 27.9, 22.5, 14.0; HRMS calcd for $C_{10}H_{22}O_4N$ (M + NH₄) 220.1549, found 220.1552.

General Procedure for Ring Contraction of the Epoxy Alcohols 1–9, 24, and 25. The epoxy alcohol was dissolved in dry toluene, and dry LiBr and dry TMU were added under stirring (magnet). The mixture was refluxed for 10 min and then cooled to room temperature. TLC analysis showed a UV-active aldehyde spot with an R_f value 2–3 times greater than the R_f values of the byproducts (the TLC plates were developed by the anisaldehyde/ H_2SO_4 reagent 16). The mixture was added to a short (10–15 cm) SiO_2 column and chromatographed to give pure 10-16 and 26+27. The eluent was chosen to give an R_f value of approximately 0.3 for the aldehyde product. The ring contractions that produced aldehydes in less than 20% yield generally appeared as brown—black mixtures, whereas reaction mixtures giving more than 20% yield were less colored; mixtures giving 10-13 and 26+27 had a weak yellow color.

(-)-(2*S*,5*S*)-2-[[[Dimethyl(1,1,2-trimethylpropyl)-silyl]oxy]methyl]-5-methoxy-2,5-dihydrofuran-3-carbaldehyde (10). Epoxy alcohol 1⁷ (145 mg, 0.45 mmol), toluene (4 mL), LiBr (69 mg, 0.79 mmol), and TMU (0.128 mL, 1.07 mmol) were treated according to the general procedure. The

⁽¹⁶⁾ Casey, M.; Leonard, J.; Lygo, B.; Procter, G. Advanced Practical Organic Chemistry, 1st ed.; Chapman & Hall: New York, 1990.

^{(17) (}a) Keck, G.; Enholm, E.; Yates, J.; Wiley, M. *Tetrahedron* **1985**, 41, 4079. (b) Keck, G.; Yates, J. *J. Am. Chem. Soc.* **1982**, 104, 5829.

10 (78 mg, 57%); $[\alpha]_D$ and NMR data were as reported. (-)-(2*R*,5*S*)-2-Butyl-5-methoxy-2,5-dihydrofuran-3-carbaldehyde (11). Epoxy alcohol 2 (171 mg, 0.84 mmol), toluene (8 mL), LiBr (124 mg, 1.43 mmol), and TMU (0.240 mL, 2.00 mmol) were treated according to the general procedure. The mixture was chromatographed (heptane/EtOAc, 4:1) to give 11 (87 mg, 56%): $[\alpha]^{23}_D - 19$ (c 2.3, CHCl₃); 1 H NMR data (CDCl₃) δ 9.90 (s, 1 H), 6.66 (dd, 1 H, J = 1.6, 1.8 Hz), 5.88 (dd, 1 H, J = 1.3, 4.2 Hz), 5.14 (m, 1 H), 3.43 (s, 3 H), 1.92 (dddd, 1 H, J = 3.2, 4.8, 10.9, 13.8 Hz), 1.59 (m, 1 H), 1.38 – 1.21 (m, 4 H), 0.89 (t, 3 H, J = 7.1 Hz); 13 C NMR data (CDCl₃) δ 187.7, 148.6, 142.3, 107.7, 83.8, 55.2, 33.5, 27.2, 23.0, 14.4; HRMS calcd for $C_{10}H_{17}O_3$ (M + H) 185.1178, found

(-)-(2*R*,5*S*)-2-(But-3-en-1-yl)-5-methoxy-2,5-dihydrofuran-3-carbaldehyde (12). Epoxy alcohol 3¹² (180 mg, 0.90 mmol), toluene (8 mL), LiBr (132 mg, 1.51 mmol), and TMU (0.250 mL, 2.08 mmol) were treated according to the general procedure. The mixture was chromatographed (heptane/ EtOAc, 4:1) to give **12** (75 mg, 46%): $[\alpha]^{23}_D - 26$ (c 2.0, CHCl₃); ¹H NMR data (CDCl₃) δ 9.90 (s, 1 H, CHO), 6.67 (dd, 1 H, J =1.4, 2.0 Hz, H-4), 5.89 (dd, 1 H, J = 1.3, 4.2 Hz, H-5), 5.81 (dddd, 1 H, J = 5.9, 6.7, 10.4, 17.1 Hz, H-3'), 5.16 (dddd, 1 H,J = 2.2, 4.3, 6.6, 9.2 Hz, H-2), 5.02 (ddd, 1 H, J = 1.6, 3.3,17.1 Hz, H-4'), 4.96 (dddd, 1 H, J = 1.2, 1.8, 3.3, 10.2 Hz, H-4'), 3.43 (s, 3 H, OMe), 2.18-2.00 (m, 3 H, H-1' and 2'), 1.72-1.61 (m, 1 H, H-1'); 13 C NMR data (CDCl₃): δ 187.7 (CHO), 148.4 (C-3), 142.4 (C-4), 138.3 (C-3'), 115.4 (C-4'), 107.7 (C-5), 83.2 (C-2), 55.2 (OMe), 32.9 (C-1'), 29.3 (C-2'); HRMS calcd for $C_{10}H_{15}O_3$ (M + H) 183.1021, found 183.0985.

(+)-(2*S*,5*S*)-2-[(Decylthio)methyl]-5-methoxy-2,5-dihydrofuran-3-carbaldehyde (13). Epoxy alcohol 4¹¹ (504 mg, 1.51 mmol), toluene (12 mL), LiBr (215 mg, 2.47 mmol), and TMU (0.425 mL, 3.54 mmol) were treated according to the general procedure. The mixture was chromatographed (heptane/EtOAc, 6:1) to give 13 (215 mg, 45%) as an oil: $[α]^{23}_D$ +4.2 (c 1.3, CHCl₃); ¹H NMR data (CDCl₃) δ 9.91 (d, 1 H, J = 0.3 Hz), 6.77 (ddd, 1 H, J = 0.3, 1.3, 1.8 Hz), 5.95 (dd, 1 H, J = 1.3, 4.1 Hz), 5.38 (dddd, 1 H, J = 2.0, 2.9, 4.1, 5.0 Hz), 3.45 (s, 3 H), 3.13 (ddd, 1 H, J = 0.4, 3.2, 13.9 Hz), 2.89 (dd, 1 H, J = 4.9, 13.9 Hz), 2.53 (t, 2 H, J = 7.5 Hz), 1.56 (m, 2 H), 1.40–1.24 (m, 14 H), 0.89 (t, 3 H, J = 6.7 Hz); ¹³C NMR data (CDCl₃) δ 187.5, 146.7, 143.1, 108.3, 83.5, 55.5, 36.0, 33.9, 32.3, 30.2, 30.0, 29.9, 29.8, 29.7, 29.6, 29.3, 14.5; HRMS calcd for C₁₇H₃₀O₃S (M⁺) 314.1916, found 314.1906.

(-)-(2R,5S)-2-[3-(Acetoxymethyl)but-3-en-1-yl]-5-methoxy-2,5-dihydrofuran-3-carbaldehyde (14). Epoxy alcohol 5¹² (237 mg, 0.87 mmol), toluene (8 mL), LiBr (123 mg, 1.42 mmol), and TMU (0.250 mL, 2.08 mmol) were treated according to the general procedure. The mixture was chromatographed (heptane/EtOAc, 3:1) to give 14 (85 mg, 38%): $[\alpha]^{23}$ _D -18 (c 0.9, CHCl₃); ¹H NMR data (CDCl₃) δ 9.90 (d, 1 H, J =0.3 Hz, CHO), 6.68 (dd, 1 H, J = 1.3, 2.1 Hz, H-4), 5.88 (dd, 1 Hz)H, J = 1.3, 4.2 Hz, H-5), 5.14 (m, 1 H, H-2), 5.04 (dd, 1 H, J= 0.8, 1.2 Hz, H-4', 4.96 (t, 1 H, J = 0.6 Hz, H-4', 4.53, 4.50(AB q, 2 H, J = 14.3 Hz, CH_2OAc), 3.42 (s, 3 H, OMe), 2.17-2.04 (m, 3 H, H-1' and 2'), 2.09 (s, 3 H, Ac), 1.77-1.65 (m, 1 H, H-1'); 13 C NMR data (CDCl₃) δ 187.6 (CHO), 171.2 (Ac), 148.2 (C-3), 143.5 (C-3'), 142.5 (C-4), 113.2 (C-4'), 107.7 (C-5), 83.1 (C-2), 67.3 (CH₂OAc), 55.2 (OMe), 31.8 (C-1'), 28.7 (C-2'), 21.4 (Ac); HRMS calcd for $C_{13}H_{18}O_5$ (M⁺) 254.1154, found 254.1153.

(-)-(2*R*,5*S*)-2-[3-(Ethoxycarbonyl)but-3-en-1-yl]-5-methoxy-2,5-dihydrofuran-3-carbaldehyde (15) and (-)-(2*S*,5*R*)-5-[3-(Ethoxycarbonyl)but-3-en-1-yl]-2-methoxy-2,5-dihydrofuran-3-carbaldehyde (15a). Epoxy alcohol 6^{12} (58 mg, 0.21 mmol), toluene (2.5 mL), LiBr (31 mg, 0.35 mmol), and TMU (0.060 mL, 0.50 mmol) were treated according to the general procedure. The mixture was chromatographed (heptane/EtOAc, 1:1) to give 15, containing 3% of its isomer 15a (20 mg, 38%); $[\alpha]_D^{23}$ -16 (*c* 2.0, CDCl₃). Compound 15: 1 H NMR data (CDCl₃) δ 9.91 (s, 1 H), 6.68 (dd, 1 H, J = 1.3, 2.0 Hz), 6.15 (m, 1 H), 5.89 (dd, 1 H, J = 1.3, 4.1 Hz), 5.54 (dd, 1 H, J = 1.4, 2.8 Hz), 5.17 (m, 1 H), 4.20 (q, 2 H, J = 7.1 Hz), 3.43 (s, 3 H), 2.34 (dd, 2 H, J = 7.5, 8.0 Hz), 2.13 (m, 1 H),

1.77 (m, 1 H), 1.31 (t, 3 H, J = 7.1 Hz); ¹³C NMR data (CDCl₃) δ 187.2, 167.1, 147.7, 142.1, 140.1, 124.7, 107.4, 82.6, 60.7, 54.9, 31.9, 26.9, 14.2; HRMS calcd for C₁₃H₁₉O₅ (M + H) 255.1232, found 255.1216. Compound **15a**: ¹H NMR data (CDCl₃) δ 9.90 (d, 1 H, J = <1 Hz), 6.67 (dd, 1 H, J = 1.5, 3.0 Hz), 6.18 (d, 1 H, J = 1.5 Hz), 5.87 (dd, 1 H, J = 1.2, 4.2 Hz), 5.49 (m, 1 H), 5.14 (m, 1 H), 4.18 (q, 2 H, J = 7.1 Hz), 1.30 (t, 3 H, J = 7.1 Hz).

(+)-(2S,5S)-2-[(Decylsulfonyl)methyl]-5-methoxy-2,5dihydrofuran-3-carbaldehyde (16). Epoxy alcohol 7¹¹ (272 mg, 0.75 mmol), toluene (6 mL), LiBr (110 mg, 1.26 mmol), and TMU (0.215 mL, 1.79 mmol) were treated according to the general procedure. The mixture was chromatographed (heptane/EtOAc, $1:1 \rightarrow 0:1$) to give **16** (23 mg, 9%), which crystallized upon standing: mp 54–56 °C; $[\alpha]^{23}_D$ +46 (c 1.4, CHCl₃); ¹H NMR data (CDCl₃): δ 9.92 (d, 1 H, J = 0.4 Hz, CHO), 6.79 (dd, 1 H, J = 1.3, 2.3 Hz, H-4), 5.96 (dd, 1 H, J =1.3, 4.0 Hz, H-5), 5.53 (dddt, 1 H, J = 0.4, 2.3, 4.0, 8.7 Hz, H-2), 3.66 (dd, 1 H, J = 2.3, 14.9 Hz, H-1'), 3.46 (s, 3 H, OMe), 3.16 (dd, 1 H, J = 8.7, 14.9 Hz, H-1'), 3.12 (dd, 2 H, J = 7.9,8.3 Hz, H-3'), 1.94-1.76 (m, 2 H, H-4'), 1.48-1.41 (m, 2 H, H-5'), 1.35-1.27 (m, 12 H), 0.89 (t, 3 H, J = 6.7 Hz, H-12'); ¹³C NMR data (CDCl₃) δ 186.6, 145.1, 142.1, 108.1, 78.2, 55.4, 55.1, 31.8, 29.5, 29.3, 29.1, 28.5, 22.7, 21.9, 14.1; HRMS calcd for C₁₇H₃₀O₅S (M⁺) 346.1814, found 346.1800.

C-Propyl 2,3,4,6-Tetra-O-acetyl-α-D-glucopyranoside (18). Compound 17^{14} (112 mg, 0.30 mmol) was dissolved in EtOAc/toluene (10 mL, 1:1) and the mixture was hydrogenated $(H_2,\,1\ atm,\,Pd/C,\,10\ mg).$ After 5 h, the catalyst was filtered off (Celite), and the solvent was removed. The residue was chromatographed (SiO₂, heptane/EtOAc, 3:1), and the crude material was crystallized from ether by addition of heptane to give pure **18** (107 mg, 95%): mp 132–133 °C; $[\alpha]^{23}_D$ +70 (c0.8, CHCl₃); ¹H NMR data (CDCl₃) δ 5.33 (t, 1 H, J = 9.2 Hz, H-3), 5.08 (dd, 1 H, J = 5.8, 9.6 Hz, H-2), 4.99 (dd, 1 H, J =9.0, 9.4 Hz, H-4), 4.25 (dd, 1 H, J = 5.2, 12.1 Hz, H-6), 4.19 (ddd, 1 H, J = 3.3, 5.8, 11.5 Hz, H-1), 4.08 (dd, 1 H, J = 2.6,12.1 Hz, H-6), 3.82 (ddd, 1 H, J = 2.6, 5.2, 9.3 Hz, H-5), 2.10, 2.06, 2.043, 2.036 (4s, 3 H each, Ac), 1.78 (m, 1 H, H-1'), 1.51-1.42 (m, 2 H, H-1',2'), 1.33 (m, 1 H, H-2'), 0.97 (t, 3 H, J = 7.2Hz, H-3'); 13 C NMR data (CDCl₃) δ 171.1, 170.6, 170.1, 170.0, 72.9, 70.9, 69.4, 68.9, 62.8, 27.7, 21.2, 21.09, 21.06, 18.6, 14.2; HRMS calcd for $C_{17}H_{27}O_9$ (M + H) 375.1655, found 375.1656.

C-Propyl 4,6-O-Benzylidene-α-D-glucopyranoside (19). Compound 18 (59 mg, 0.16 mmol) was dissolved in MeOH (2 mL), and methanolic NaOMe (0.015 mL, 0.5 M) was added. After 3 h, SiO₂ (ca 0.5 g) was added, and the mixture was filtered (Celite) and concentrated. The residue was suspended in a mixture of CH₃CN (2.0 mL, freshly distilled) and α,α dimethoxytoluene (0.10 mL, 0.67 mmol). p-TsOH (catalytic amount) was added, and the mixture was stirred for 24 h. The acid was neutralized by passing the reaction mixture through a short pad of Al₂O₃ (basic, grade II). The solvent was removed to give **19** (41 mg, 88%), which crystallized upon standing: mp 195–197 °C; $[\alpha]^{23}_D$ +51 (c 0.6, CHCl₃); ¹H NMR data (CDCl₃) δ 7.51 (m, 2 H, Ph), 7.39 (m, 3 H, Ph), 5.55 (s, 1 H, PhCH), 4.28 (dd, 1 H, J = 4.6, 10.1 Hz, H-6), 4.10 (m, 1 H, H-1), 3.90 (m, 2 H, H-2,3), 3.71 (t, 1 H, J = 10.1 Hz, H-6), 3.62 (dt, 1 H, J = 4.6, 9.7 Hz, H--5, 3.47 (t, 1 H, J = 9.3 Hz, H--4), 2.65 (d, 1 H, J = 1.8 Hz, OH), 2.44 (d, 1 H, J = 2.4 Hz, OH), 1.71 (m, 2 H, H-1'), 1.53 (m, 1 H, H-2'), 1.36 (m, 1 H, H-2'), 0.99 (t, 3 H, J = 7.3 Hz, H-3'); ¹³C NMR data (CDCl₃) δ 137.5, 129.8, 128.8, 126.7, 102.4, 82.6, 76.8, 72.8, 72.1, 69.9, 63.7, 27.0, 19.0, 14.3; HRMS calcd for $C_{16}H_{23}O_5$ (M + H) 295.1545, found 295.1549.

C-Propyl 4,6-O-Benzylidene-2,3-anhydro-α-D-allopyranoside (20). Compound 19 (243 mg, 0.82 mmol) was dissolved in toluene (9 mL), and K_2CO_3 (1.32 g, 9.55 mmol), powdered NaOH (480 mg, 12.0 mmol), 4-toluenesulfonyl chloride (409 mg, 2.14 mmol), and DMSO (0.40 mL) were added. The mixture was stirred at room temperature for 12 h and at 55 °C for 10 h and then cooled to room temperature. 2-Methoxyethanol (0.45 mL, 5.70 mmol) and tetrabutylammonium hydrogen sulfate (76.2 mg, 0.22 mmol) were added, and the stirring was continued for 28 h at room temperature. The mixture was poured into water/toluene (50 mL, 1:1), and the organic phase was washed with water (3 × 10 mL), dried (Na₂-

SO₄), and concentrated to give a crude ditosylate (497 mg). Attempts to purify the ditosylate on silica were unsuccessful. The crude ditosylate (396 mg, approx. 0.66 mmol) was dissolved in a mixture of toluene (7 mL), DMSO (0.35 mL) and 2-methoxyethanol (0.15 mL), and aqueous sodium hydroxide (2 mL, 50%) and tetrabutylammonium hydrogen sulfate (76.2 mg) were added. The reaction mixture was stirred at room temperature for 8 days, then refluxed for 17 h, and treated as in the preparation of the mixture of 21 and 20 below. The crude product was chromatographed (SiO2, heptane/EtOAc, 4:1) to give **20** (34 mg, 18%): mp 90-92 °C; $[\alpha]^{23}_D$ +40 (c 0.9, CHCl₃); ¹H NMR data (CDCl₃) δ 7.53 (m, 2 H, Ph), 7.36–7.40 (m, 3 H, Ph), 5.58 (s, 1 H, PhCH), 4.20 (ddd, 1 H, J = 0.7, 4.9, 10.2 Hz, H-6), 4.07 (ddd, 1 H, J = 3.2, 5.8, 8.6 Hz, H-1), 4.01 (dd, 1 H, J = 1.2, 9.0 Hz, H-4), 3.85 (ddd, 1 H, J = 4.9, 9.0,10.2 Hz, H-5), 3.65 (t, 1 H, J = 10.2 Hz, H-6), 3.58 (br d, 1 H, J = 4.6 Hz, H-3), 3.40 (dd, 1 H, J = 3.2, 4.6 Hz, H-2), 1.83 (m, 1 H, H-1'), 1.70 (m, 1 H, H-1'), 1.40-1.55 (m, 2 H, H-2'), 1.00 (t, 3 H, J = 7.3 Hz, H-3'); ¹³C NMR data (CDCl₃) δ 137.7, 129.6, 128.8, 126.7, 103.1, 78.8, 71.4, 69.8, 61.4, 56.1, 52.2, 32.3, 19.2, 14.4; HRMS calcd for $C_{16}H_{21}O_4$ (M + H) 277.1440; found 277.1444.

C-Propyl 4,6-O-Benzylidene-2,3-anhydro-α-D-mannopyranoside (21) and C-Propyl 4,6-O-Benzylidene-2,3-anhydro-α-D-allopyranoside (20). Compound 19 (769 mg, 2.61 mmol) was dissolved in dry pyridine (75 mL), and 4-toluenesulfonyl chloride (1030 mg, 5.40 mmol) was added. The mixture was left at room temperature for 1 h, then heated at 60 °C for 24 h, and cooled to room temperature. The solvent was removed by coevaporation with two portions of toluene, the residue was partitioned between ether and water, and the aqueous phase was extracted with ether (5 \times 50 mL). The extract was dried (Na₂SO₄), filtered, and concentrated. The syrupy residue was chromatographed (SiO₂, heptane/EtOAc, 3:1) to give a regioisomeric mixture of monotosylates (971 mg, 83%). Part of the mixture (237 mg, 0.53 mmol) was dissolved in MeOH (5 mL), and methanolic NaOMe (2 mL, 0.5 M) was added under stirring. An additional portion of methanolic NaOMe (0.5 mL) was added after 3 days, and the stirring was continued for 24 h. CH₂Cl₂ (25 mL) was added, and the mixture was washed with water (3 × 5 mL), dried (Na₂SO₄). filtered, and concentrated. The residue was chromatographed (heptane/EtOAc, 3:1) to give 20 (25 mg, 25%) and 21 (95 mg, 65%). Both **20** and **21** crystallized upon standing. Compound **21**: mp 113–115 °C; $[\alpha]^{23}_D$ +28 (c 0.8, CHCl₃); ¹H NMR data (CDCl₃) δ 7.52 (m, 2 H, Ph), 7.39 (m, 3 H, Ph), 5.60 (s, 1 H, PhCH), 4.25 (dd, 1 H, J = 4.7, 10.4 Hz, H-6), 4.18 (dd, 1 H, J = 4.4, 10.1 Hz, H-1), 3.72 (d, 1 H, J = 10.4 Hz, H-6), 3.68 (d, 1 H, J = 9.6 Hz, H-4), 3.50 (d, 1 H, J = 3.8 Hz, H-3), 3.35 (dt, 1 H, J = 4.7, 9.8 Hz, H-5), 3.07 (d, 1 H, J = 3.8 Hz, H-2), 1.85-1.94 (m, 1 H, H-1'), 1.44-1.61 (m, 3 H, H-1',2'), 1.01 (t, 3 H, J = 7.2 Hz, H-3'); 13 C NMR data (CDCl₃) δ 137.6, 129.7, 128.8, 126.6, 102.7, 76.2, 72.8, 70.3, 63.3, 54.5, 53.2, 32.2, 19.5, 14.2; HRMS calcd for $C_{16}H_{21}O_4$ (M + H) 277.1440, found 277.1437.

C-Propyl 2,3-Anhydro-α-D-allopyranoside (22). Compound 20 (149 mg, 0.54 mmol) was dissolved in THF/EtOH (4.5 mL, 3:1), and the mixture was hydrogenated (H₂, 1 atm, Pd/C, 38 mg). The reaction was monitored by TLC (heptane/ EtOAc, 2:1). The starting material was consumed after 6 days. and the catalyst was filtered off (Celite). The solvent was removed, and the residue was chromatographed (toluene/ EtOAc, 1:1 \rightarrow 0:1) to give **22** (75 mg, 74%) as an oil: $[\alpha]^{23}$ _D +26 (c 1.6, CHCl₃); ¹H NMR data (CDCl₃) δ 4.03 (ddd, 1 H, J = 3.5, 6.0, 9.0 Hz, H-1, 3.95 (br t, 1 H, J = 7.9 Hz, H-4, 3.76,3.71 (br ABq, 2 H, J = 11.7 Hz, H-6), 3.52 (dd, 1 H, J = 1.8, 4.5 Hz, H-3), 3.47 (dd, 1 H, J = 3.6, 4.5 Hz, H-2), 3.43 (m, 1 H, H-5), 3.26 (d, 1 H, J = 8.2 Hz, OH-4), 2.69 (m, 1 H, OH-6), 1.77-1.68 (m, 1 H, H-1'), 1.65-1.56 (m, 1 H, H-1'), 1.51-1.35 (m, 2 H, H-2'), 0.96 (t, 3 H, J = 7.3 Hz, H-3'); ¹³C NMR data (CDCl₃) δ 70.5, 70.4, 66.5, 62.9, 58.5, 55.5, 31.5, 19.2, 14.4; HRMS calcd for C₉H₁₇O₄ (M + H) 189.1127, found 189.1127.

C-Propyl 2,3-Anhydro-α-D-mannopyranoside (23). Compound 21 (571 mg, 2.07 mmol) was dissolved in THF/EtOH (13.5 mL, 3:1), and the mixture was hydrogenated (H_2 , 1 atm, Pd/C, 102 mg). The reaction was monitored by TLC (heptane/EtOAc, 2:1). The starting material was consumed after 10

days, and the catalyst was filtered off (Celite). The solvent was removed, and the residue was chromatographed (toluene/EtOAc, 1:1 \rightarrow 0:1) to give **23** (293 mg, 75%), which crystallized from an ether/heptane mixture: mp 75–78 °C; $[\alpha]^{23}_{\rm D}$ +5.3 (c 1.4, CDCl₃); $^1{\rm H}$ NMR data (CDCl₃) δ 4.11 (dd, 1 H, J=4.3, 10.0 Hz, H-1), 3.84 (br d, 1 H, J=8.8 Hz, H-4), 3.74 (m, 2 H, H-6), 3.50 (br d, 1 H, J=2.3 Hz, OH), 3.31 (d, 1 H, J=3.8 Hz, H-2 or 3), 3.21 (ddd, 1 H, J=4.3, 4.5, 9.0 Hz, H-5), 3.06 (d, 1 H, J=3.8 Hz, H-2 or 3), 2.70 (br s, 1 H, OH), 1.84 (m, 1 H, H-1'), 1.40–1.55 (m, 3 H, H-1'/2'), 0.98 (t, 3 H, J=7.2 Hz, H-3'); $^{13}{\rm C}$ NMR data (CDCl₃) δ 71.3, 70.2, 63.5, 63.2, 56.1, 53.3, 31.3, 19.4, 14.3; HRMS calcd for ${\rm C_9H_{17}O_4}$ (M + H) 189.1127, found 189.1123.

C-Propyl 2,3-Anhydro-6-O-[dimethyl(1,1,2-trimethylpropyl)silyl]-α-D-allopyranoside (24). Compound 22 (67) mg, 0.36 mmol) was dissolved in dry pyridine (5 mL), and dimethyl(1,1,2-trimethylpropyl)chlorosilane (0.085 mL, 0.43 mmol) was added at room temperature under stirring. After 2 days, a second portion of dimethyl(1,1,2-trimethylpropyl-)chlorosilane (0.007 mL, 0.03 mmol) was added, and the mixture was stirred at -18 °C overnight. TLC (toluene/EtOAc, 1:1) showed that 22 had been consumed. MeOH (1 drop) and toluene (3 × 10 mL) were added, and the solvents were removed. The syrupy residue was dissolved in ether (75 mL), and the organic phase was washed with water (3 \times 5 mL), dried (Na₂SO₄), and concentrated. The residue was chromatographed (CH₂Cl₂/EtOAc, $100:1 \rightarrow 10:1$) to give **24** (19 mg, 16%) as a syrup: $[\alpha]^{23}_D$ -3.2 (c 1.6, CHCl₃); ¹H NMR data (CDCl₃) δ 4.00 (ddd, 1 H, J = 3.4, 5.8, 8.9 Hz, H-1), 3.93 (ddd, 1 H, J= 2.1, 5.2, 7.7 Hz, H-4), 3.76 (dd, 1 H, J = 5.3, 10.2 Hz, H-6),3.67 (dd, 1 H, J = 6.3, 10.2 Hz, H-6), 3.51 (dd, 1 H, J = 2.1, 4.4 Hz, H-3), 3.47 (dt, 1 H, J = 5.8, 8.2 Hz, H-5), 3.43 (dd, 1 Hz)H, J = 3.4, 4.4 Hz, H-2), 3.00 (br d, 1 H, J = 5.2 Hz, OH-4), 1.73 (m, 1 H, H-1'), 1.63 (m, 2 H, H-1' and Me₂CH), 1.45 (m, 2 H, H-2'), 0.97 (t, 3 H, J = 7.3 Hz, H-3'), 0.87 (d, 6 H, J = 6.9Hz, Me₂CH), 0.85 (s, 6 H, Me₂C), 0.13 (s, 6 H, Me₂Si); ¹³C NMR data (CDCl₃) δ 70.2, 69.7, 69.2, 65.5, 57.7, 54.8, 34.5, 31.6, 25.5, 20.7, 20.6, 19.1, 18.9, 18.8, 14.4, -3.1, -3.2; HRMS calcd for C₁₇H₃₅O₄Si (M + H) 331.2305, found 331.2307.

C-Propyl 2,3-Anhydro-6-O-[dimethyl(1,1,2-trimethylpropyl)silyl]-α-D-mannopyranoside (25). Compound 23 (266 mg, 1.41 mmol) was dissolved in dry pyridine (20 mL), and dimethyl(1,1,2-trimethylpropyl)chlorosilane (0.335 mL, 1.70 mmol) was added at room temperature under stirring. After 18 h, TLC (toluene/EtOAc, 1:1) showed that 23 had been consumed. MeOH (1 drop) and toluene (3 \times 25 mL) were added, and the solvents were removed. The syrupy residue was dissolved in ether (200 mL), and the organic phase was washed with water (3 × 20 mL), dried (Na₂SO₄), and concentrated. The residue was chromatographed (CH₂Cl₂/EtOAc, 10: 1) to give 25 (285 mg, 61%) as a syrup [chromatography with heptane/EtOAc, 4:1, gave a mixture of 25 and the isomer silylated in the 4-position (407 mg, 10:1)]. Compound 25: $[\alpha]^{23}_{D}$ –19 (c 0.9, CHCl₃); ¹HNMR data (CDCl₃) δ 4.08 (dd, 1 H, J = 4.4, 10.2 Hz, H-1), 3.81 (dd, 1 H, J = 5.3, 9.5 Hz, H-6), 3.76 (dd, 1 H, J = 1.3, 8.6 Hz, H-4), 3.57 (dd, 1 H, J = 9.0, 9.5Hz, H-6), 3.28 (d, 1 H, J = 1.7 Hz, OH-4), 3.26 (d, 1 H, J = 3.9Hz, H-2 or -3), 3.23 (m, 1 H, H-5), 3.03 (d, 1 H, J = 3.8 Hz, H-2 or -3), 1.85 (m, 1 H, H-1'), 1.61 (heptet, 1 H, J = 6.9 Hz, Me₂CH), 1.39–1.56 (m, 3 H, H-1',2'), 0.98 (t, 3 H, J= 7.1 Hz, H-3'), 0.88 (d, 6 H, J = 6.8 Hz, Me₂CH), 0.86 (s, 6 H, Me₂C), 0.14 (s, 6 H, Me₂Si); 13 C NMR data (CDCl₃) δ 71.3, 68.7, 67.6, $66.6,\ 55.1,\ 53.0,\ 34.5,\ 31.3,\ 25.6,\ 20.7,\ 20.5,\ 19.4,\ 18.9,\ 18.8,$ 14.2, -3.2, -3.3; HRMS calcd for $C_{17}H_{35}O_4Si$ (M + H) 331.2305, found 331.2312.

(2*R*,5*S*)-2-Propyl-5-[[[Dimethyl(1,1,2-trimethylpropyl)silylloxy|methyl]-2,5-dihydrofuran-3-carbaldehyde (26). (a) Epoxy alcohol 24 (16.3 mg, 0.049 mmol), toluene (0.8 mL), LiBr (8.1 mg, 0.093 mmol), and TMU (0.015 mL, 0.125 mmol) were treated according to the general procedure. The mixture was chromatographed (heptane/EtOAc, 10:1) to give a mixture (81:18:1 according to ¹H-NMR and GC-MS analysis: DB-Wax 30 m capillary column, 100 °C for 3 min, then 3 °C/min) of 26, 27, and 28 (9.3 mg, 61%).

(b) Epoxy alcohol **25** (201 mg, 0.61 mmol), toluene (6 mL), LiBr (99 mg, 1.14 mmol), and TMU (0.200 mL, 1.67 mmol)

were treated according to the general procedure. The mixture was chromatographed (heptane/EtOAc, 10:1) to give a mixture (71:17:12 according to ¹H-NMR and GC-MS analysis) of **26**, **27**, and **28** (114 mg, 60%). A sample of **26** (90–95% purity) was obtained by preparative GC (6 m OV-351 column with inner diameter 4 mm; column temperature 175 °C; injector temperature 225 °C; detector temperature 210 °C; injected volume of **26** dissolved in ether: 0.150 mL, 100 mg/mL; elution time 225-230 min). Compounds 27 and 28 were not obtained pure enough for full structural analysis; ¹H NMR and HRMS of the crude material was consistent with the structures given. Compound **26**: ¹H NMR data (CDCl₃) δ 9.83 (s, 1 H, CHO), 6.90 (t, 1 H, J = 1.7 Hz, H-4), 5.11 (dddd, 1 H, J = 1.7, 4.2, 5.7, 6.3 Hz, H-2), 5.01 (dddd, 1 H, J = 1.7, 3.5, 5.1, 8.8 Hz, H-5), 3.81 (dd, 1 H, J = 4.2, 10.2 Hz, CH_2OSi), 3.64 (dd, 1 H, J = 6.3, 10.2 Hz, CH₂OSi), 1.83 (m, 1 H, OCHC H_2 CH₂), 1.66-1.54 (m, 2 H, OCHCH2CH2, Me2CH), 1.45-1.33 (m, 2 H, OCHCH₂C H_2), 0.93 (t, 3 H, J = 7.3 Hz, CH₂C H_3), 0.87 (d, 6 H, J = 6.8 Hz, Me_2 CH), 0.84 (s, 6 H, Me_2 C), 0.10 (s, 6 H, Me_2 Si); 13 C NMR data (CDCl₃) δ 187.4 (CHO), 148.5 (C-4), 146.1 (C-3), 86.1 (C-5), 84.5 (C-2), 65.4 (C-1"), 36.8 (C-1"), 34.6 (Me_2 CH), 25.5 (Me_2 C), 20.7, 18.9, 18.6, 14.4 (C-3"), -3.1 (Me_2 Si); HRMS calcd for $C_{17}H_{33}O_3$ Si (M + H): 313.2199, found 313.2186.

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Supporting Information Available: ¹H NMR spectra for all title compounds described in the Experimental Section (16 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

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